

# Evaluation of the use of lignite of Turkey's with biomass as Agricultural waste as fuel in terms of emissions

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**Abstract.** Agricultural wastes as Biomass contains low carbon, high hydrogen, high oxygen and a lower amount of sulfur. Coals contain higher amounts of carbon, lower amounts of hydrogen, lower oxygen and higher amounts of sulfur. With the use of lignite and biomass mixture as fuel will provide less CO<sub>2</sub> and SO<sub>2</sub> emissions and a more economical mixture will be obtained. Considering these emissions, fluidized bed combustion systems are recommended in the literature for the combustion of lignite and biomass. In this study, rice husks, corn cobs, walnut shells, sunflower shells, olive cake and woodchips were used as agricultural waste. 10 different lignite extracted from Turkey were used as fuel. It has been assumed that the combustion process was carried out by taking the biomass rate of 10%, 30% and 50%. When burning of 1 kg of lignite and biomass mixture, the highest CO<sub>2</sub> emission occurs from 10% woodchips - 90% Kütahya - Ömerler (washed) mixture as 2.938 kg and the highest SO<sub>2</sub> emission obtained from 10% olive cake - 90% Kütahya Seyitömer-Ayvalı lignite mixture as 0.061 kg. The highest H<sub>2</sub>O emission was obtained by mixing 50% woodchips - 50% Manisa-Kısrakdere lignite as 0.563 kg.

## 1 Introduction

Most of the energy needs that increase with population and industrialization are met from fossil fuels. Environmental pollution caused by the production and use of energy from fossil fuels and the limited reserves of fossil resources is the most important problems encountered today. Today, it is important for energy policies to develop appropriate technologies for the utilization of low-quality lignite. Biomass is a renewable fuel type and its pollutant properties are low compared to coal. In Turkey, which is an agricultural country, biomass, one of the renewable energy sources, has a wide usage potential. Combustion of coal and biomass by mixing, economical use of low-quality coal and biomass are important in terms of reducing the consumption of both produced pollutants and fossil fuels. It is necessary to examine the combustion behaviour of coal biomass mixtures of these systems. Thus, biomass and coal have different burning behaviours due to their chemical composition. The combination of fossil fuels such as coal and biomass creates an opportunity for the development of new technologies and a great economic contribution [1, 2].

A. Elorf and B. Sarh examined the effect of  $\lambda = 1.3, 1.7, 2.3,$  and  $2.7$  different excess air ratios on the pulverized olive cake (OC) of combustion dynamics in a vertical oven [1]. D. Kopuz investigated the Tekirdağ Malkara and Çorum Bayat lignites and peach kernel and

cherry kernel biomasses also, the combustion behavior of their mixtures in different proportions [2]. B. Engin et al have presented experimental studies related to the combustion of 18 lignite samples obtained from various lignite fields of Turkey in a 30 kWth circulating fluidized bed combustor [3]. L. Armestoa et al have demonstrated the importance of burning coal and foot cake together in terms of utilizing a very specific biomass waste from the olive oil industry. Two different types of coal, lignite, and anthracite, were used for the study. Flammability tests were carried out in a fluidized bed facility [4]. W. K. Kordylewski et al have researched slag reduction by mixing Polish lignite and bituminous coal with biomass as laboratory-scale preliminary. Three agricultural biomass fuels (cereal straw sunflower husk, rape straw) and olive cake were used in the experiments [5]. T. Madhiyanon et al have investigated the Combustion properties of rice husk in combination with bituminous coal, emissions of fuel mixtures, and combustion efficiency were investigated together with the 120 kWth cyclone fluidized-bed combustor [6]. A. T. Atımtay and M. Varol examined the combustion performances and emission characteristics of olive cake and coal mixture (Tuncbilek lignite) in bubbling fluidized bed. Burning tests were carried out with olive cake mixing ratios of 25%, 50%, and 75% by weight [7]. S. V. Vassilev et al in their study, extensive research has done on the phase-mineral transformations of organic and inorganic matter that occur during biomass combustion [8]. M. Varol et al investigated the combustion of Bursa-Orhaneli lignite and

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10%, 30%, and 50% wood chips mixtures with circulating fluidized bed combustor with excess air ratio varying between 1.05-2.43 [9]. M. A. Bennini et al examined the burning of raw olive pomace, exhausted olive pomace and deoiled olive pomace in a fixed bed boiler in their study [10]. A. A. Bhuiyan and J. Naser investigated the co-combustion of biomass coal mixture in a 0.5MW small-scale furnace under air and an oxy-fuel condition was studied. The necessary programs for chemical reactions, heat transfer, fluid and particle flow fields and turbulence calculation are coded [11]. A. Pimchuai et al investigated the torrefaction process of rice husk and mixtures of four different agricultural wastes such as sawdust, peanut husks, bagasse, and water hyacinth with lignite [12]. D. Vamvuka and S. Saxioni were examined the composition of ash produced from fixed bed combustion experiments using various urban waste and their mixtures with lignite were investigated. Slagging and fouling trends and environmental effects on recycling to soil have been studied. The effects of the combustion temperature and the percentage of biomass in the mixture were examined [13]. W. Jerzak et al examined the combustion of cedarnut hull and high sulfur lignite mixture in a fluidized bed reactor. In the study, it has been shown that burning bituminous coal with cedarnut shells under the same conditions does not show a significant decrease in SO<sub>2</sub> emissions [14]. A. Elorf et al investigated the effect of inlet vortex motion on the flow behavior and combustion dynamics of the olive cake (OC) milled in a 3-dimensional vertical furnace [15]. N. Vardar and N. İlten measured the flue gas emissions generated in the facilities where the pomace is burned in Balıkesir province, which has approximately 12-25% of the olive pomace production in Turkey and the emissions were evaluated according to the oxygen in the flue gas [16]. W. Jerzak et al investigated the combustion of nutshells and bituminous coal in a new two-disc reactor supporting two fluidized beds stacked one above the other [17].

The purpose of this study is to obtain a new lignite-biomass fuel mixture using 10 different lignites with the highest ratio of carbon which obtained from different regions of Turkey and 6 different biomass. Then the CO<sub>2</sub>, SO<sub>2</sub> and H<sub>2</sub>O emission amounts that will occur by burning these fuel mixtures in fluidized bed boilers were calculated. The ratio of agricultural wastes used as biomass has been accepted as 10%, 30% and 50%. The combustion process was assumed to be full combustion and air excess coefficient was 1.5.

## 2 Methodology

### 2.1 Chemical formulas of fuel mixes

The total mass of K mixtures ( $m_{mixT}$ ) is the sum of the individual masses of the mixes. The total mole amount of the mixture is ( $n_{mixT}$ ) the sum of the individual molar amounts of the mixtures [18].

$$m_{mixT} = \sum_{i=1}^K m_i \quad (1)$$

$$n_{mixT} = \sum_{i=1}^K n_i \quad (2)$$

The ratio of the individual molar amounts of the mixture to the molar amount of the total mixture ( $y_{ni}$ ) is the molar ratio.

$$y_{ni} = \frac{n_i}{n_T} \quad (3)$$

The ratio of the individual mass amounts of the mixture to the mass amount of the total mixture is the mass ratio ( $y_{mi}$ ) is the mass ratio.

$$y_{mi} = \frac{m_i}{m_T} \quad (4)$$

Sum of the mass ratio ( $y_{mi}$ ) and mole ratio ( $y_{ni}$ ) is one.

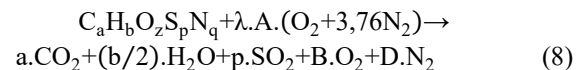
$$\sum_{i=1}^K y_{mi} = 1 \quad (5)$$

$$\sum_{i=1}^K y_{ni} = 1 \quad (6)$$

The molar quantity ( $n$ ) is equal to the divided by the molar mass ( $M$ ) of a substance mass ( $m$ ) [18].

$$n = \frac{m}{M} \quad (7)$$

### 2.2 Combustion equations and emission calculation



If oxygen is balanced for A, B and D, the following equations are obtained.  $\lambda$  is excess air ratio [19].

$$A = (a + \frac{b}{4} + p - \frac{z}{2}) \quad (9)$$

$$B = (\lambda - 1) \cdot (a + \frac{b}{4} + p - \frac{z}{2}) \quad (10)$$

$$D = 3,76 \cdot \lambda \cdot (a + \frac{b}{4} + p - \frac{z}{2}) + \frac{q}{2} \quad (11)$$

$$M_{CO_2} = \frac{a \cdot CO_2}{M} = \frac{a \cdot 44}{M} \quad \text{kgCO}_2 \quad (12)$$

$$M_{SO_2} = \frac{p \cdot SO_2}{M} = \frac{p \cdot 64}{M} \quad \text{kgSO}_2 \quad (13)$$

$$M_{H_2O} = \frac{b \cdot H_2O}{M} = \frac{b \cdot 9}{M} \quad \text{kgH}_2\text{O} \quad (14)$$

### 2.3 Properties of fuels

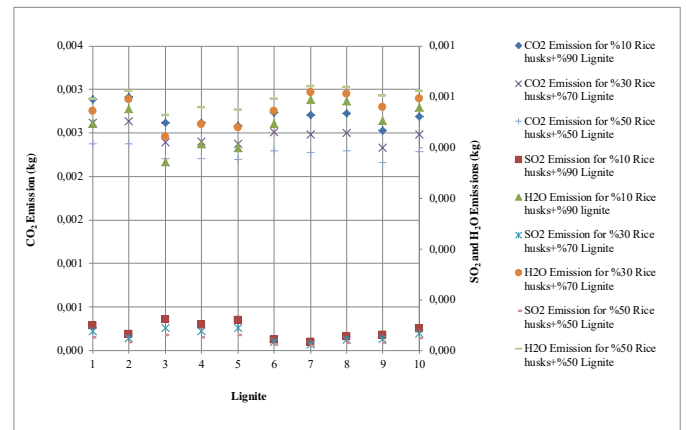
Table 1 shows the lignite types used in the study and chemical ratios (wt %) of various agricultural wastes as biomass .

**Table 1.** The chemical content (wt%) of different lignite and biomass [3, 7-9]

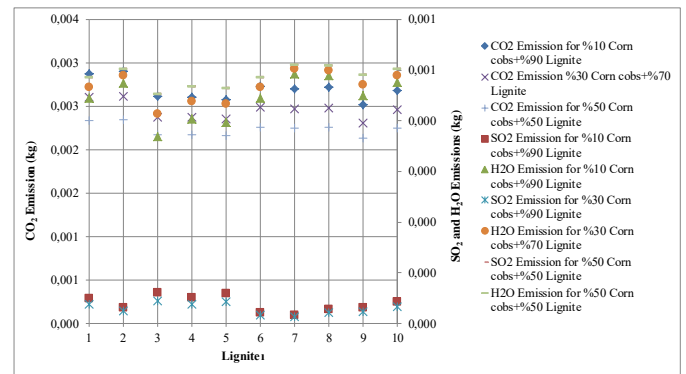
Fuel	Lignite				
	Chemical content				
	C	H	N	O	S
Kütahya-Tunçbilek (washed)	70.58	4.12	2.19	6.23	2.39
Kütahya-Ömerler (washed)	69.30	4.31	2.44	5.48	1.49
Kütahya-Seyitömer-Ayvalı	53.70	2.74	1.12	11.66	2.51
Kütahya-Seyitömer-Höyükaltı	56.35	3.24	1.18	12.63	2.19
Kütahya-Seyitömer (washed)	55.36	3.15	1.22	12.93	2.55
Manisa İmbat (washed)	66.54	4.11	1.98	11.71	1.01
Manisa Kırakdere	66.55	4.69	1.06	13.44	0.79
Manisa Dereköy	67.55	4.68	1.28	12.23	1.34
Muğla Eskihisar	54.00	3.67	1.30	15.03	1.31
Bursa Orhaneli	69.26	4.74	0.97	13.50	2.18
Biomass					
Rice husks	49.3	6.1	0.8	43.6	0.08
Corn cobs	47.8	5.9	0.5	45.6	0.01
Walnut shells	49.8	6.2	1.4	42.4	0.09
Sunflower shells	50.4	5.5	1.1	42.9	0.03
olive cake	51.4	5.8	1.3	36.9	0.08
Woodchips	53.7	6.9	0.3	37.7	0.00

### 3 Results and Discussion

In Figure 1, the change of CO<sub>2</sub>, SO<sub>2</sub> and H<sub>2</sub>O emissions for a) rice husks b) corn cobs biomass mixed with different lignites at different rates are given. Figure 2 shows the CO<sub>2</sub>, SO<sub>2</sub> and H<sub>2</sub>O emissions changes for a) walnut shells b) sunflower shells biomass mixed with different lignites at different rates. Figure 3 shows the CO<sub>2</sub>, SO<sub>2</sub> and H<sub>2</sub>O emissions for a) olive cake b) woodchips biomass mixed with different lignites at different rates.

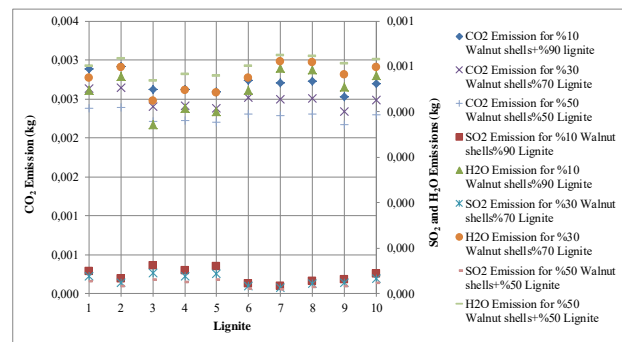


(a)

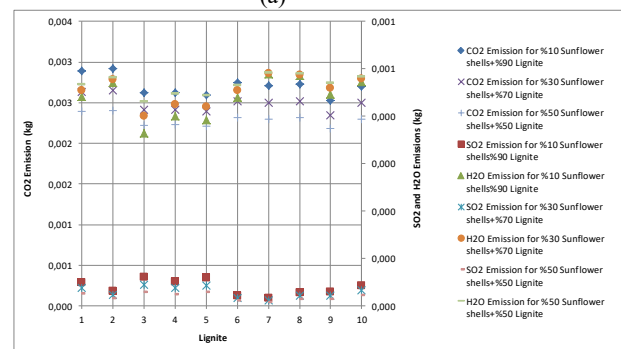


(b)

**Fig. 1.** The change of CO<sub>2</sub>, SO<sub>2</sub> and H<sub>2</sub>O emissions for (a) rice husks (b) corn cobs biomass mixed with different lignites at different rates

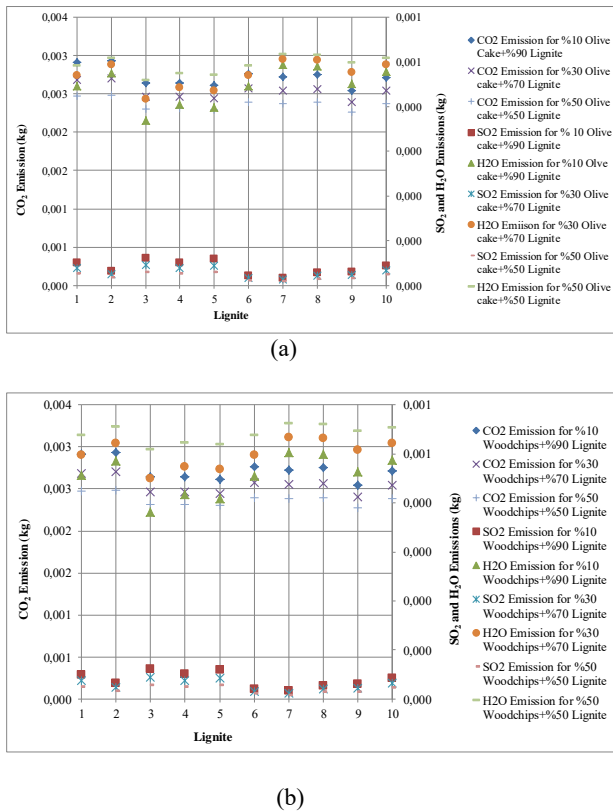


(a)



(b)

**Fig. 2.** The CO<sub>2</sub>, SO<sub>2</sub> and H<sub>2</sub>O emissions changes for (a) walnut shells (b) sunflower shells biomass mixed with different lignites at different rates



**Fig. 3.** The change of CO<sub>2</sub>, SO<sub>2</sub> and H<sub>2</sub>O emissions for (a) olive cake (b) woodships biomass mixed with different lignites at different rates

By burning 1 kg of 10% rice husks and 90% lignite composition, the highest CO<sub>2</sub> emission with 2.913 kg occurs in Kütahya-Ömerler (washed) and the lowest in Muğla-Eskihisar lignite with 2.524 kg. The highest SO<sub>2</sub> emission occurs in Kütahya Seyitömer-Ayvalı with 0.060 kg, while the lowest occurs in Manisa Kırskdere lignite with 0.016 kg. The highest H<sub>2</sub>O emission is 0.495 kg in Manisa Kırskdere lignite, while Kütahya Seyitömer-Ayvalı with the lowest 0.371 kg occurs. With the burning of 1 kg of 30% rice husks and 70% lignite composition, the highest CO<sub>2</sub> emission with 2.635 kg occurs in Kütahya-Ömerler (washed) and the lowest in Muğla-Eskihisar lignite with 2.332 kg. The highest SO<sub>2</sub> emission occurs in Kütahya Seyitömer-Ayvalı with 0.044 kg, while the lowest occurs in Manisa Kırskdere lignite with 0.013 kg. With the burning of 1 kg of 50% rice husks and 50% lignite composition, the highest CO<sub>2</sub> emission with 2.378 kg occurs in Kütahya-Ömerler (washed) and the lowest in Muğla-Eskihisar lignite with 2.162 kg. The highest SO<sub>2</sub> emission occurs in Kütahya Seyitömer-Ayvalı with 0.030 kg, while the lowest in Manisa Kırskdere lignite with 0.009 kg. The highest H<sub>2</sub>O emission is 0.521 kg in Manisa Kırskdere lignite and Kütahya Seyitömer-Ayvalı with the lowest 0.464 kg.

By burning 1 kg of 10% corn cobs and 90% lignite composition, the highest CO<sub>2</sub> emission with 2.907 kg occurs in Kütahya-Ömerler (washed) and the lowest in Muğla-Eskihisar lignite with 2.517 kg. The highest SO<sub>2</sub> emission occurs in Kütahya Seyitömer-Ayvalı with 0.061 kg, while the lowest in Manisa Kırskdere lignite with 0.016 kg. The highest H<sub>2</sub>O emission occurs in the Manisa

Kırskdere lignite with 0.493 kg and in Kütahya Seyitömer-Ayvalı lignite with the lowest 0.369 kg. With the burning of 1 kg of 30% corn cobs and 70% lignite composition, the highest CO<sub>2</sub> emission with 2.617 kg occurs in Kütahya-Ömerler (washed) and the lowest in Muğla-Eskihisar lignite with 2.313 kg. The highest SO<sub>2</sub> emission occurs in Kütahya Seyitömer-Ayvalı with 0.044 kg, while the lowest occurs in Manisa Kırskdere lignite with 0.012 kg. The highest H<sub>2</sub>O emission is 0.502 kg in Manisa Kırskdere lignite and Kütahya Seyitömer-Ayvalı with the lowest 0.414 kg. By burning 1 kg of 50% corn cobs and 50% lignite composition, the highest CO<sub>2</sub> emission occurs in Kütahya-Ömerler (washed) with 2,348 kg, and the lowest in Muğla-Eskihisar lignite with 2,162 kg. The highest SO<sub>2</sub> emission occurs in Kütahya Seyitömer-Ayvalı with 0.029 kg, while the lowest in Manisa Kırskdere lignite with 0.009 kg. The highest H<sub>2</sub>O emission occurs in Manisa Kırskdere lignite with the highest 0.521 kg and Kütahya Seyitömer-Ayvalı with the lowest 0.511 kg.

With the burning of 1 kg of 10% walnut shells and 90% lignite composition, the highest CO<sub>2</sub> emission with 2.915 kg occurs in Kütahya-Ömerler (washed) and the lowest in Muğla-Eskihisar lignite with 2.526 kg. The highest SO<sub>2</sub> emission occurs in Kütahya Seyitömer-Ayvalı with 0.061 kg, while the lowest in Manisa Kırskdere lignite with 0.016 kg. The highest H<sub>2</sub>O emission occurs in Manisa Kırskdere lignite with 0.496 kg and lowest in Kütahya Seyitömer-Ayvalı lignite with 0.373 kg. By burning 1 kg of 30% walnut shells and 70% lignite composition, the highest CO<sub>2</sub> emission with 2.641 kg occurs in Kütahya-Ömerler (washed) and the lowest in Muğla-Eskihisar lignite with 2.339 kg. The highest SO<sub>2</sub> emission occurs in Kütahya Seyitömer-Ayvalı with 0.045 kg, while the lowest in Manisa Kırskdere lignite with 0.013 kg. The highest H<sub>2</sub>O emission occurs in Manisa Kırskdere lignite with 0.511 kg and Kütahya Seyitömer-Ayvalı lignite with lowest 0.424 kg. With the burning of 1 kg of 50% walnut shells and 50% lignite composition, the highest CO<sub>2</sub> emission with 2.388 kg occurs in Kütahya-Ömerler (washed) and the lowest in Muğla-Eskihisar lignite with 2.172 kg. The highest SO<sub>2</sub> emission occurs in Kütahya Seyitömer-Ayvalı with 0.030 kg, while the lowest in Manisa Kırskdere lignite with 0.009 kg. The highest H<sub>2</sub>O emission occurs in Manisa Kırskdere lignite with the highest 0.526 kg and Kütahya Seyitömer-Ayvalı with the lowest 0.469 kg.

By burning 1 kg of 10% sunflower shells and 90% lignite composition, the highest CO<sub>2</sub> emission with 2.918 kg occurs in Kütahya-Ömerler (washed) and the lowest in Muğla-Eskihisar lignite with 2.529 kg. The highest SO<sub>2</sub> emission occurs in Kütahya Seyitömer-Ayvalı with 0.061 kg, while the lowest in Manisa Kırskdere lignite with 0.016 kg. The highest H<sub>2</sub>O emission occurs in Manisa Kırskdere lignite with 0.489 kg and Kütahya Seyitömer-Ayvalı lignite with the lowest 0.364 kg. With the burning of 1 kg of 30% sunflower shells and 70% lignite composition, the highest CO<sub>2</sub> emission with 2.648 kg occurs in Kütahya-Ömerler (washed) and the lowest in Muğla-Eskihisar lignite with 2.346 kg. The highest SO<sub>2</sub> emission occurs in Kütahya Seyitömer-Ayvalı with 0.044 kg, while the lowest in Manisa Kırskdere lignite with



0.012 kg. The highest H<sub>2</sub>O emission occurs in Manisa Kısırakdere lignite with 0.490 kg, Kütahya Seyitömer-Ayvalı with the lowest 0.400 kg. By burning 1 kg of 50% sunflower shells and 50% lignite composition, the highest CO<sub>2</sub> emission with 2.399 kg occurs in Kütahya-Ömerler (washed) and the lowest in Muğla-Eskihisar lignite with 2.184 kg. The highest SO<sub>2</sub> emission occurs in Kütahya Seyitömer-Ayvalı with 0.030 kg, while the lowest in Manisa Kısırakdere lignite with 0.009 kg. The highest H<sub>2</sub>O emission occurs in Manisa Kısırakdere lignite with 0.492 kg and Kütahya Seyitömer-Ayvalı with the lowest 0.432 kg.

With the burning of 1 kg of 10% olive cake and 90% lignite composition, the highest CO<sub>2</sub> emission with 2.937 kg occurs in Kütahya-Ömerler (washed) and the lowest in Muğla-Eskihisar lignite with 2.548 kg. The highest SO<sub>2</sub> emission occurs in Kütahya Seyitömer-Ayvalı with 0.061 kg, while the lowest in Manisa Kısırakdere lignite with 0.016 kg. The highest H<sub>2</sub>O emission is 0.494 kg in Manisa Kısırakdere lignite and Kütahya Seyitömer-Ayvalı with the lowest 0.370 kg. With the burning of 1 kg of 30% olive cake and 70% lignite composition, the highest CO<sub>2</sub> emission with 2.701 kg occurs in Kütahya-Ömerler (washed) and the lowest in Muğla-Eskihisar lignite with 2.398 kg. The highest SO<sub>2</sub> emission occurs in Kütahya Seyitömer-Ayvalı with 0.045 kg, while the lowest in Manisa Kısırakdere lignite with 0.013 kg. The highest H<sub>2</sub>O emission is 0.507 kg in Manisa Kısırakdere lignite and Kütahya Seyitömer-Ayvalı with the lowest 0.418 kg. By burning 1 kg of 50% olive cake and 50% lignite composition, the highest CO<sub>2</sub> emission with 2.479 kg occurs in Kütahya-Ömerler (washed), and the lowest in Muğla-Eskihisar lignite with 2.263 kg. The highest SO<sub>2</sub> emission occurs in Kütahya Seyitömer-Ayvalı with 0.031 kg, while the lowest in Manisa Kısırakdere lignite with 0.010 kg. The highest H<sub>2</sub>O emission occurs in Manisa Kısırakdere lignite with a maximum of 0.519 kg and in Kütahya Seyitömer-Ayvalı lignite with a minimum of 0.460 kg.

By burning 1 kg of 10% woodchips and 90% lignite composition, the highest CO<sub>2</sub> emission with 2.938 kg occurs in Kütahya-Ömerler (washed) and the lowest in Muğla-Eskihisar lignite with 2.549 kg. The highest SO<sub>2</sub> emission occurs in Kütahya Seyitömer-Ayvalı with 0.061 kg, while the lowest in Manisa Kısırakdere lignite with 0.016 kg. The highest H<sub>2</sub>O emission is 0.504 kg in Manisa Kısırakdere lignite and Kütahya Seyitömer-Ayvalı with the lowest 0.382 kg. With the burning of 1 kg of 30% woodchips and 70% lignite composition, the highest CO<sub>2</sub> emission with 2.702 kg occurs in Kütahya-Ömerler (washed) and the lowest in Muğla-Eskihisar lignite with 2.402 kg. The highest SO<sub>2</sub> emission occurs in Kütahya Seyitömer-Ayvalı with 0.044 kg, while the lowest in Manisa Kısırakdere lignite with 0.012 kg. The highest H<sub>2</sub>O emission is 0.534 kg in Manisa Kısırakdere lignite and Kütahya Seyitömer-Ayvalı with the lowest 0.450 kg. With the burning of 1 kg of 50% woodchips and 50% lignite composition, the highest CO<sub>2</sub> emission with 2.483 kg occurs in Kütahya-Ömerler (washed) and the lowest in Muğla-Eskihisar lignite with 2.271 kg. The highest SO<sub>2</sub> emission occurs in Kütahya Seyitömer-Ayvalı with 0.029 kg, while the lowest in Manisa Kısırakdere lignite with

0.009 kg. The highest H<sub>2</sub>O emission occurs in Manisa Kısırakdere lignite with the highest 0.563 kg and lowest 0.509 kg in Kütahya Seyitömer-Ayvalı lignite.

The most important chemical elements in terms of fuel properties and emissions are carbon, hydrogen, oxygen and sulfur. The carbon percentage in 10 different lignites selected is over 50%. Kütahya Tunçbilek lignite has the highest carbon rate. The carbon ratio of this lignite is 70.58%. The lowest is Kütahya Seyitömer-Ayvalı lignite. Its carbon ratio is 53.70%. The highest hydrogen rate is 4.74% in Bursa Orhaneli lignite. The lowest is 2.74% of lignite extracted from Kütahya Seyitömer-Ayvalı. The highest oxygen content is 15.03% of lignite extracted from Muğla Eskihisar. The lowest is 5.48% of the lignite mined by Kütahya Ömerler. The highest sulfur content is Kütahya Seyitömer (washed) lignite and it is 2.55%. The highest is Manisa Kısırakdere lignite and its ratio is 0.79%.

Woodchips have the highest carbon ratio among agricultural wastes selected as Biomass with 53.7%. The lowest is corn cobs with 47.8%. The highest hydrogen content is woodchips with 6.9%, while the lowest is 5.5% sunflower shells. The highest oxygen rate is 45.6% with corn cobs and the lowest is olive cake with 36.9%. The highest sulfur rate is Walnut shells with 0.09%, while the agricultural waste that does not contain any sulfur is woodchips.

## 4 Conclusions

As a result, the highest CO<sub>2</sub> emission was obtained in Kütahya-Ömerler (Washed) and the lowest in Muğla-Eskihisar lignite by burning 1 kg of fuel mixture in a fluidized bed combustion system obtained by combining all types of lignite and all biomass agricultural wastes in certain proportions. The highest SO<sub>2</sub> emission was obtained in Kütahya-seyitömer-Ayvalı region lignite and the lowest in Manisa-Kısırakdere lignites. H<sub>2</sub>O emission was obtained in Manisa-Kısırakdere lignite and lowest in Kütahya Seyitömer-Ayvalı lignite.

The determinant of CO<sub>2</sub> and SO<sub>2</sub> emissions in all of the fuel mixtures obtained by combining different types of lignite and agricultural wastes in certain proportions is the carbon and sulfur ratios in the lignite used. Lignite, which has the highest carbon and sulfur ratio, is decisive in the CO<sub>2</sub> and SO<sub>2</sub> emissions in the new fuel mixture. The decisive factor in H<sub>2</sub>O emission is the hydrogen ratio in agricultural wastes. In the mixture of lignite and agricultural waste with the highest hydrogen ratio, H<sub>2</sub>O emission occurs in the highest amount.

As the biomass rate increases, while CO<sub>2</sub> and SO<sub>2</sub> emissions decrease, H<sub>2</sub>O emissions increase. This is because, according to Biomass, while the carbon ratio and sulfur ratio in lignite is higher, the hydrogen ratio is less. Since the hydrogen rate in biomass is higher than lignite, the higher the biomass rate, the more H<sub>2</sub>O is obtained as a result of the burning event. On the other hand, less CO<sub>2</sub> and SO<sub>2</sub> emissions occur.

For the %10 biomass - %90 lignite mixtures, by burning of 1 kg lignite and biomass mixture, the highest CO<sub>2</sub> emission occurs from 10% woodchips - 90% Kütahya - Ömerler (washed) as 2.938 kg and the highest

SO<sub>2</sub> emission obtained from 10% olive cake - 90% Kütahya Seyitömer-Ayvalı lignite mixture as 0.061 kg. The highest H<sub>2</sub>O emission is obtained by mixing 10% woodchips - 90% Manisa-Kısrakdere lignite as 0.504 kg.

For the %30 biomass - % 70 lignite mixtures, with the burning of 1 kg lignite and biomass mixture, the highest CO<sub>2</sub> emission occurs from 30% woodchips - 70% Kütahya - Ömerler (washed) as 2.702 kg and the highest SO<sub>2</sub> emission obtained from 30% olive cake - 70% in Kütahya Seyitömer-Ayvalı lignite mixture as 0.045 kg. The highest H<sub>2</sub>O emission is obtained by mixing 30% woodchips - 70% Manisa-Kısrakdere lignite as 0.534 kg.

For the %50 biomass - %50 lignite mixtures, when 1 kg lignite and biomass mixture is burned, the highest CO<sub>2</sub> emission occurs from 50% woodchips - 50% Kütahya - Ömerler (washed) with 2.483 kg and the highest SO<sub>2</sub> emission obtained from 50% olive cake - 50% in Kütahya Seyitömer-Ayvalı lignite mixture as 0.031 kg. The highest H<sub>2</sub>O emission is obtained by mixing 50% woodchips - 50% Manisa-Kısrakdere lignite as 0.563 kg.

In the literature, the combustion properties of new fuels which limited number formed by the mixture of different lignite types and agricultural wastes are examined. In this study, agricultural wastes and lignites located in Turkey, emissions by burning new fuels that great number formed by mixing in different proportions is examined. Thus, it was possible to compare new fuels. Accordingly, the better mixing ratio and fuel type can be determined. In later studies, the combustion properties and emission amounts of the fuels that will be formed by mixing different types of biomass such as garbage waste, waste paper, grape waste and sewage sludge with lignite will be investigated.

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